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**DIURNAL VARIATIONS OF DISSOLVED OXYGEN, BIOCHEMICAL  
AND CHEMICAL OXYGEN DEMANDS IN THE WATER SAMPLES OF  
SHANTHISAGAR LAKE, DAVANGERE DISTRICT, KARNATAKA**

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**ABSTRACT**

The present study is carried out to determine the diurnal variation of water quality parameter in Shanthisagar lake, Davangere district. In this study, various physico-chemical parameters like temperature, pH, dissolved oxygen, biological oxygen demand and sediment oxygen demand were determined. By considering the physio-chemical parameters of water obtained in the present study the dissolved oxygen is rich in water is beneficial for aquatic life. The physico-chemical parameters showed the diverse variation in the lake.

**Keywords: Diurnal variations, DO, BOD, SOD, Shanthisagar lake, water quality**

**1. INTRODUCTION**

Freshwater bodies are considered as one of the important natural wealth for the survival of all the living things of the biosphere. The rate of deterioration of water quality of fresh water resources like lentic and lotic water

bodies is now a worldwide problem. For sustainable utilization of the water resources, periodic assessment of the freshwater bodies are very much necessary [1]. Nutrient rich water bodies are often associated with an increase in algal blooms

and hypoxia in aquatic systems indicates eutrophic status [2-3].

Lakes are considered to be one of the most productive and biologically rich inland surface water ecosystems. Lake is defined as an aggregate of water bodies They are either natural or man made water bodies. They are either natural or man-made. It is estimated that India has about 4.1 million It is estimated that, India has about 4.1 million hectares of wet land of which 1.5 million hectares are natural and 2.6 million hectares are man made. A study by the Wild Life Institute of India reveals that, 70-80 % of individual fresh water marshes and Lakes have been lost during last 50 years.

There are three major lakes around Davangere city. Due to rapid developments and heavy urbanization there is a reduction in catchment area. In addition to this, other human activities and discharge of sewage have caused heavy pollution to these Lakes Hence, these Lakes are at the verge of these Lakes. Hence these lakes are at the verge of disappearance. If these Lakes are properly restored it will greatly help in solving the water problems along with other environmental advantages. Keeping this in view, the present work is undertaken and as a case study.

Ecological conditions may change significantly within aquatic ecosystem during 24 hours period. Solar radiation varies from high intensities at mid-day to darkness. Surface temperature and concentrations of dissolved gases may also fluctuate between extremes of day and night, especially in shallow, littoral areas [16-19].

Dissolved oxygen is an indication of the general health of the water bodies. The objectives of the study is to determine the diurnal variation of dissolved oxygen, biological oxygen demand and sediment oxygen demand in Shanthisagar lake. However, the study of diurnal variations in water with respect to Shanthi Sagar lake in particular is lacking. Hence, the present investigation is carried out.

## **2. MATERIALS AND METHOD**

### **2.1 Study Area**

Shanthi Sagara lake is the second largest irrigation water body in Asia. It is found in Kerebilchi village of Channagiri taluk of Davanagere district in Karnataka, India (**Figure 1& 2**).

Shanti Sagara lake, has a water spread area of about 6,550 acres (2,651 ha), has a circumference of 30 km. It has a total drainage basin of 81,483 acres.

It irrigates 4,700 acres of land and more than 50 villages are benefited by it.

The lake receives the drainage of 20 square miles. The embankment is constructed between two hills, and embankment is of no great length; it is around 950 ft, but it is of stupendous width (Max 120 ft (37 m), min 70 ft (21 m)), height and strength, though not quite straight. The main road connecting between Channagiri and Davanagere pass through on this embankment. It has resisted successfully the floods of centuries, but owing to the great pressure of the volume of the water in lake (Mysore: a gazetteer, vol 2, p. 482).

## 2.2 Analysis

Present study was conducted during January 2015. Samples were collected every day in the morning and afternoon during every day. Stopped BOD glass bottles used for water sample collection were cleaned with distilled water prior to sample collection. Containers were again rinsed with sampling water before collecting water sample. Temperature was recorded in the field with the help of mercury thermometer and dissolved oxygen samples were fixed with alkaline potassium iodide and manganoous sulphate at sampling sites. Sediment samples were collected manually by adding some sediments from the bottom to it. The collected samples were brought to

the laboratory for analysis. The analysis is done by modified winkler's method [4, 14].



Figure 1: Location of the study area (Source:www.veethi.com)



Figure 2: Views of Shanthisagar lake

## 3. RESULTS AND DISCUSSION

### 3.1 Variation of DO and BOD in water sample of Shanthisagar lake.

Figure 4 and 5 shows the diurnal variation of DO and BOD during morning and afternoon. The DO in Shanthisagar lake varied from 3.8 to 10.6 mg/L. The BOD in lake varied from 6.1 to 8.1 mg/L. The DO is high and BOD is low which is favorable for aquatic organisms.

### 3.2 Diurnal variation of DO and SOD

Figure 6 shows the diurnal variation of DO and SOD. The DO in this water body varied from 3.7 to 4.8 mg/L. The SOD in Shanthisagar lake varied from 19.5 to 28.6 mg/L.

### 3.3 Water analysis

The difference between air and water temperatures was never more than 4°C in Shanthisagar lake but Aravinda *et al* [8] have never noticed more than 1.8°C in Kundawada lake of Davangere district. Most of the time sunlight was available when an over saturation in DO of the lake water was recorded. Afternoon onwards DO concentration gradually minimum. The pH of the lake water was alkaline and it ranged from 7.9 to 8.5 (**Figure 3**). It was found to be directly related to changes in CO<sub>3</sub> concentration as pH of the water increased during the morning time and decreased in the afternoon. The increase in pH during day time may be due to high rate of photosynthesis, although the range of variation of pH was usually minimum.

Water transparency is an important factor in the growth and distribution of living beings [7]. The transparency has indirect relationship with nutrients and was affected by the growth of algal biomass. Transparency depicted an inverse relationship with a number of water quality parameters in diel variations.

Total alkalinity of water is due to the presence of mineral salts and it was primarily caused by the CO<sub>3</sub> and HCO<sub>3</sub> ions. High alkalinity leads to the increase in temperature which also coincides with the nutrients and

HCO<sub>3</sub> in particular. It has also been observed that relatively higher values of alkalinity were favorable for the growth of plankton as reported by many researchers [7, 9-10].

Total dissolved solids (TDS) of water indicates the density of chemicals and as a measure of edaphic relationship that contributes to the yield of water [11]. The chloride content of water gave an idea of organic matter, nitrate and harmful substances [12, 7]. Dissolved oxygen (DO) is importance in the study of status of a lake. DO content has direct influence on primary productivity of water. BOD is defined as the amount of oxygen required by the microbes to stabilizing decomposable organic matter under aerobic condition. COD indicates the oxygen required for chemical oxidation of organic matter and oxidisable inorganic contents with the help of strong chemical oxidants [7].

Harish Kumar *et al* [13] studied the seasonal variations in the ionic levels such as phosphate, nitrate, carbonate, bicarbonate and chloride in the surface water samples of Kundavada lake of Davangere district (Karnataka) of the study area caused by natural and human anthropogenic activities. Their results showed that the lake is included under mesotrophic category and needs certain degree of

treatment and it also protected from the possible source of contamination. The water quality parameters have been variously engaged in the classification of water bodies and the level of nutrient enrichments [15, 19].

Anusha Balangoda [20] evaluated the diurnal and seasonal Dissolved Oxygen fluctuations and their effects on water quality, was conducted in a eutrophic polymictic reservoir. She reported that artificial aeration increased the dissolved oxygen concentration in the deep portion of the reservoir and the existing aeration was inadequate to amplify the DO concentration in the entire reservoir.

### 3.4 Correlation coefficient

There is a little variations in water temperature than air temperature is a major and critical limiting factor for biotic components because most of the aquatic organisms often have a narrow temperature tolerance [5-7]. A positive correlation was recorded between air and water temperature ( $r=0.905$ ) at 0.01 level. Water transparency has an inverse correlation with many of the water parameters. pH has positive correlation with air and water temperature and negative relation with water transparency ( $r=-0.270$ ). Dissolved oxygen showed positive correlation with water temperature, transparency and pH. The correlation of Do

with many parameters supports the role of photosynthesis in governing the above parameters.

BOD has negative relation with transparency, DO and positive correlation with temperature and pH. COD has positive relation with many parameters and negative correlation with transparency and BOD. Total alkalinity showed positive correlation with water temperature ( $r=0.930$ ), BOD ( $r=0.990$ ) and COD ( $r=0.950$ ) and negative correlation with transparency ( $r=-0.680$ ). Total hardness (TH) has positive correlation with air, water temperature, pH, COD, BOD and alkalinity. TH showed negative relation with transparency ( $r=-0.850$ ) and DO ( $r=-0.450$ ). Chloride content of water has positive correlation with water temperature, BOD, COD and total alkalinity. Chloride has negative relation with transparency and DO ( $r=-0.580$ ). Total dissolved solids (TDS) showed negative relation with transparency, DO and positive correlation with temperature, BOD, COD, total alkalinity, TH and chloride (**Table 1**).

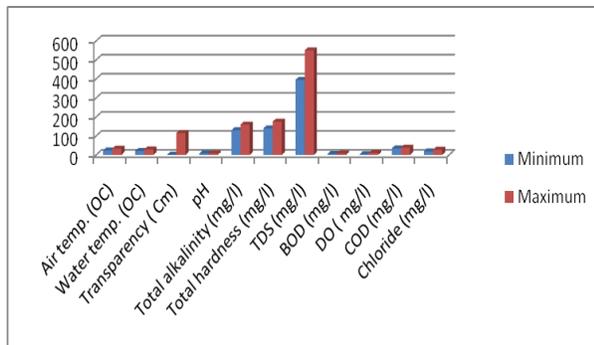


Figure 3: Range of physico-chemical parameters of water in Shanthisagar lake

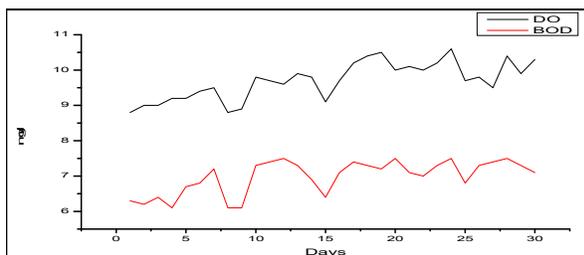


Figure 4: Diurnal variations of DO and BOD during morning

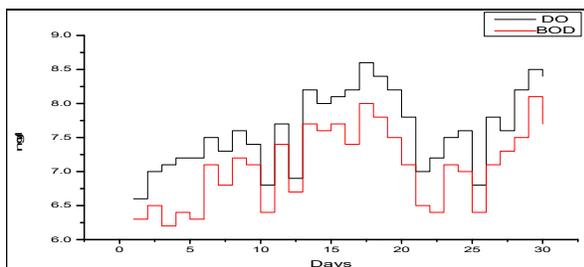


Figure 5: Diurnal variations of DO and BOD during Afternoon

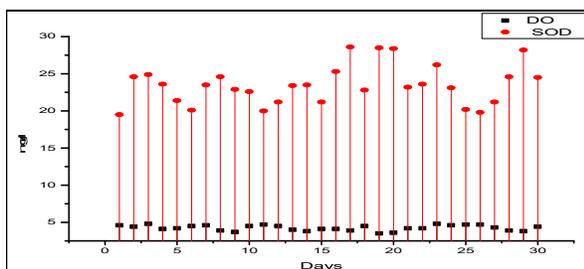


Figure 6: Diurnal variations of DO and SOD in Shanthisagar lake

#### 4. CONCLUSION

In this present study, it was found that the maximum and minimum parameters were not at the level of pollution. In the light of

standard of water quality recommended by WHO, the lake water should be used for drinking, cooking and fisheries. In order to maintain the health of the lake with respect to water quality it is essential that authorities should take immediate step on the awareness should be created regarding the effect of water pollution on the health of the human beings. People should be advised at least to boil the water to kill the pathogens before the water used for drinking purpose. The present study will be supportive for further ecological consideration and monitoring of water quality, production, fisheries improvement and lake management strategy in future. The current study will be helpful for further assessment in monitoring the water quality, yield, fishery development and lake management in future.

#### CONFLICT OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this article.

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Table 1: Correlation coefficient between different physico-chemical parameters in Shanthisagar lake, Davangere district

	AT	WT	Transp.	pH	DO	BOD	COD	TA	TH	Cl	TDS
AT	0.000										
WT	0.905	0.000									
Transp.	-0.670	-0.751	0.000								
pH	0.799	0.475	-0.270	0.000							
DO	-0.040	0.448	0.195	0.475	0.0000						
BOD	0.690	0.910	-0.870	0.180	-0.710	0.000					
COD	0.821	0.928	-0.750	0.420	-0.500	0.900	0.000				
TA	0.695	0.930	-0.680	0.178	-0.730	0.990	0.950	0.000			
TH	0.878	0.985	-0.850	0.430	-0.450	0.915	0.970	0.940	0.000		
Cl	0.750	0.910	-0.645	0.240	-0.580	0.850	0.900	0.900	0.920	0.000	
TDS	0.920	0.980	-0.705	0.510	-0.430	0.880	0.950	0.910	0.980	0.950	0.000

\*Values above 0.50 depicting significant at 0.01 level; 0.51 and 0.40 values significant at 0.05 level